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MTL TR 86-1

# **EFFECT OF OXIDATION ON** THE DENSIFICATION OF SINTERABLE RBSN

GEORGE E. GAZZA

January 1986



Prepared under Interagency Agreement DE-A105-21411

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U.S. ARMY MATERIALS TECHNOLOGY LABORATORY WATERTOWN, MASSACHUSETTS 02172-0001

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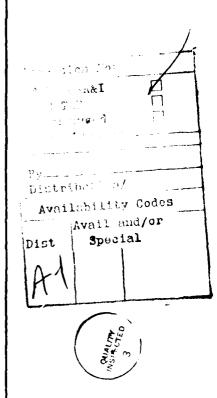
REPORT DOCUMENTATION PAGE		READ INSTRUCTIONS BEFORE COMPLETING FORM		
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4. TITLE (and Subtitle)	l	5. TYPE OF REPORT & PERIOD COVERED		
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SINTERABLE RBSN		Final Report		
STATEMENT ADDA		6 PERFORMING ORG. REPORT NUMBER		
7. AUTHOR(s)		8. CONTRACT OR GRANT NUMBER(s)		
George E. Gazza				
9. PERFORMING ORGANIZATION NAME AND ADDRESS		10. PROGRAM ELEMENT, PROJECT, TASK AREA & WORK UNIT NUMBERS		
U.S. Army Materials Technology Laboratory		Interagency Agreement		
Watertown, MA 02172-0001		DE-A105-21411		
SLCMT-MCC		Agency Accession: DA OH4676		
II. CONTROLLING OFFICE NAME AND ADDRESS		12. REPORT DATE January 1986		
U.S. Department of Energy, Office of Transportation Systems, Advanced Materials		13. NUMBER OF PAGES		
Development Program, Washington, DC 20545		8		
14. MONITORING AGENCY NAME & ADDRESS(it differen		15. SECURITY CLASS. (of this report)		
		Unclassified		
		15a, DECLASSIFICATION DOWNGRADING		
		SCHEDULE		
16. DISTRIBUTION STATEMENT (of this Report)				
Approved for public release; distribution unlimited.				
17. DISTRIBUTION STATEMENT (of the abstract entered in Block 20, if different from Report)				
TA CURRY ENERGY AND VALUE				
18. SUPPLEMENTARY NOTES Presented at the American Ceramic Society, Pacific Coast Meeting, Basic				
Science Division, San Francisco, CA, October 28-31, 1984.				
272 2772-2011, Dan 12411-2000, Oil, October 20 31, 17071				
19. KEY WORDS (Continue on reverse side if necessary and identify by block number)				
Sintering Mechanical properties				
Silicon nitrides Stress rupture				
Oxidation	Oxidation Yttria			
20. ABSTRACT (Continue on reverse side if necessery and identify by block number)				
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# ABSTRACT

Predoped RBSN bars containing 6%203 and  $1\%Fe_203$  were sintered to high density using a two-step pressure/temperature cycle. Some bars were preoxidized at  $1000^{\circ}$ C prior to sintering to increase the oxygen content ( $5i0_2$ ) of the specimens and to determine its effect on sintering behavior. Modulus of rupture and stress-rupture data were obtained on both as-received and heat-treated specimens



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#### INTRODUCTION

Sintering of silicon nitride usually involves mixing silicon nitride powder with an appropriate additive, e.g., MgO or Y<sub>2</sub>O<sub>3</sub>, cold pressing the mixture, then sintering to achieve densification. An alternate approach involves the sintering of reaction-bonded silicon nitride (RBSN) where the densification aid has been added to the silicon prior to nitridation or infiltrated into the RBSN subsequent to nitridation. A major advantage cited for the sintering of doped RBSN is the relatively low percent shrinkage (6 to 8%) during densification which produces improved dimensional control. Another difference to be noted between the sintering processes is the compositional and phase development sequence occurring prior to and during sintering. Since quaternary oxynitride compounds are formed during the nitridation step of predoped RBSN, the reaction kinetics will differ for each sintering process. Those kinetics are influenced by the amount and stability of each component or compound formed and the reaction path. Additionally, high nitrogen gas pressures and appropriate cover powders are necessary for stabilizing specimen compositions at high sintering temperatures.

In the present study, reaction-bonded silicon nitride bars that had been fabricated from injection-molded silicon mixed with 6%Y2O3, and 1%Fe2O3 were sintered to high density using N2 gas pressures up to 8.0 MPa and temperatures between 1875°C and 2000°C. The starting SiO2 content of some specimens was increased by oxidation prior to sintering. Densified specimens were tested at temperatures between 700°C and 1200°C in stress rupture for up to 1000 hours under stress (300 MPa). Room temperature modulus of rupture (RT MOR) was determined for surviving stress-rupture specimens and on specimens not subjected to high temperature testing. Fracture surfaces were examined by SEM to define fracture origins.

#### EXPERIMENTAL PROCEDURE

Injection molded, predoped RBSN bars containing 6%Y2O3 and 1%Fe2O3 were supplied by the Garrett Turbine Engine Company, Phoenix, AZ, (identified as code 9) for study. X-ray diffraction (XRD) analysis of as-received bars indicated that they were typically 80% alpha phase. The remainder was essentially beta phase and yttrium-nitrogen apatite (Y5Si3O12N) with a small amount of free silicon also detected. The densities of the specimens were between 2.4 and 2.5 g/cc. Some of the bars were oxidized at 1000°C, in air, for times ranging from 10 to 120 minutes. The increases in SiO2 content were estimated from observed weight gains. Both oxidized and unoxidized specimens were imbedded in packing powder in a covered RBSN crucibile during sintering. The composition of the packing powder was 64%Si3N4-25%BN-6%Y2O3-5%SiO2. It was used to protect the specimens from carbon in the atmosphere and provide a partial pressure of SiO around the specimens to restrict the loss of SiO<sub>2</sub> by reducing reactions cited in the technical literature. 1,2 High N<sub>2</sub> gas pressures were used to suppress the decomposition of  $Si_3N_4$  which could occur at the sintering temperatures utilized, i.e.,  $1875^{\circ}C$  to  $2000^{\circ}C$ . Two-step sintering procedures were used which involved using a relatively low N2 gas (Figure 1) pressure (2.0 MPa) during the first sintering step then raising the pressure to 5.5 MPa in the second step. Sintering temperatures ranged from 1875°C to 2000°C. temperature levels were generally used for the sintering process where the change in temperature level corresponded with changing the gas pressure level. The first

<sup>1.</sup> LANGE, F. F. Silicon Nitride Polyphase Systems: Fabrication, Microstructure, and Properties. International Metals Reviews, v. 25, no. 1, 1980, p. 1-20.

<sup>2.</sup> MESSIER, D. R., and DeGUIRE, E. J. Thermal Decomposition in the System Si-Y-Al-O-N. J. Amer. Cer. Soc., v. 67, no. 9, 1984, p. 602-605.

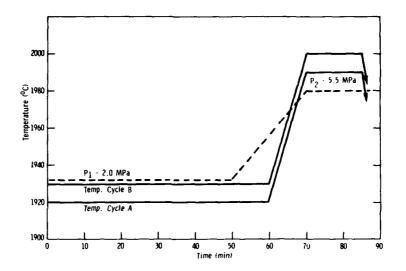


Figure 1. Two-step temperature/pressure cycles used for sintering.

temperature level employed was in the range of  $1875^{\circ}$ C to  $1940^{\circ}$ C, while the second temperature step was usually higher,  $1970^{\circ}$ C to  $2000^{\circ}$ C. Hold times of 45 to 60 minutes were used for step 1, and 15 to 30 minutes for step 2. For most sintering cycles, a crystallization treatment was performed by holding at a temperature between  $1075^{\circ}$ C and  $1200^{\circ}$ C for 60 minutes subsequent to the high temperature sintering steps. Some sintered bars were tested without machining to determine RT MOR. Bars were also machined to dimensions of 2.54 mm thick x 3.81 mm wide x 50 mm long for stress-rupture testing. The bars were placed under a 300 MPa flexure stress during the stepped temperature stress-rupture (STSR) procedure from  $800^{\circ}$ C to  $1200^{\circ}$ C at 24-hour increments. After STSR, the bars were held at  $1200^{\circ}$ C under stress for up to 1000 hours unless failure occurred. Bars surviving 750 to 1000 hours were unloaded, cooled, and removed from the furnace for determination of retained RT MOR. Similar isothermal stress-rupture experiments were also conducted at  $1000^{\circ}$ C and  $1000^{\circ}$ C to determine whether any intermediate temperature degradation occurred.

## OXIDATION OF RBSN STARTING SPECIMENS

The use of oxidation reactions to alter the SiO<sub>2</sub> content of Si<sub>3</sub>N<sub>4</sub> compositions has been previously reported by Greskovich<sup>4</sup> and Evans and Moulson.<sup>5</sup> In this study, some of the as-received, reaction-bonded Si<sub>3</sub>N<sub>4</sub>-6%Y<sub>2</sub>O<sub>3</sub>-1%Fe<sub>2</sub>O<sub>3</sub> bars were oxidized in air at  $1000^{\circ}$ C for various times up to 120 minutes before sintering. The weight changes produced by oxidation were used to estimate the increase in SiO<sub>2</sub> content. Figure 2 shows the percent weight gain for the oxidation time used at  $1000^{\circ}$ C. Parabolic kinetics are followed with little increase in SiO<sub>2</sub> content produced after approximately 60 minutes. The increase in SiO<sub>2</sub> content after 60 minutes is estimated to be 1.0 to 1.50 percent.

<sup>3.</sup> QUINN, G. D. Characterization of Turbine Ceramics After Long-Term Environmental Exposure. U.S. Army Materials Technology Laboratory, AMMRC TR 80-15, April 1980.

GRESKOVICH, C. D., and PALM, J. A. Development of High Performance Sintered Si<sub>3</sub>N<sub>4</sub>. General Electric Co., Contract DAAG46-78-C-0058, Final Report, AMMRC TR 80-46, September 1980.

<sup>5.</sup> EVANS, J. R. G., and MOULSON, A. J. The Effect of Impurities on the Densification of Reaction-Bonded Silicon Nitride (RBSN) J. of Mat. Sci., v. 18, no. 12, 1983, p. 3721-3728.

Both oxidized and as-received RBSN were densified in two-step sintering processes as described previously and illustrated in Figure 3. As shown in the figure, the oxidized bars attained higher sintered densities than those not pretreated. The best results appear to be obtained using 30 to 60 minutes oxidation time. An increase of 3 to 5% in density is noted for oxidized specimens compared with unoxidized. Weight change measurements on the sintered bars revealed that unoxidized bars always gained weight (up to 2%) during sintering while oxidized bars usually lost weight (up to 2.5%) depending on the amount of preoxidation (percent increase in SiO2 content). As the specimen becomes silica-rich, lower melting compositions may be formed. Also, preoxidation of the specimens may result in compositions where the partial pressure of SiO produced in the specimen is greater than that produced by the cover powder resulting in weight loss during sintering.

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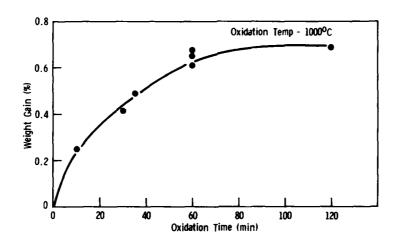


Figure 2. Effect of oxidation on as-received RBSN specimens.

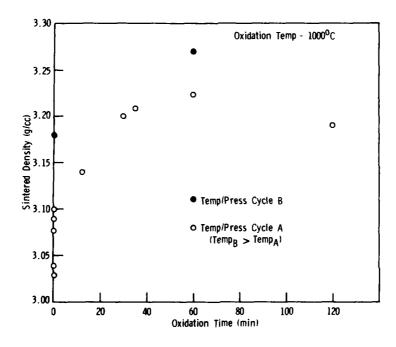


Figure 3. Effect of preoxidation time and sintering cycle on sintered density.

X-ray diffraction results on sintered specimens indicate that grain boundary phases found are similar to those in unsintered specimens that have not been pre-oxidized. The principal boundary phase appears to be  $Y_5(SiO_4)_3N$  which may remain after sintering or convert to  $Y_4_{.67}(SiO_4)_3O$ . The preoxidized specimens exhibit a change in the grain boundary phase after sintering from  $Y_5(SiO_4)_3N$  to  $Y_2SiO_5$  and occasionally some  $Y_2Si_2O_7$ . All of the  $\alpha$ -Si<sub>3</sub>N<sub>4</sub> converted to the  $\beta$  form.

# Mechanical Properties

Some bars were tested in the as-sintered condition to determine RT modulus of rupture. Values ranged from 490 to 670 MPa. Fracture origins were observed to be primarily pores/pore clusters remaining in the specimens after sintering.

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Other specimens were machined to dimensions previously cited for stress-rupture testing at temperatures of 700°C, 1000°C, and 1200°C under a 300 MPa stress. The specimens were held under these temperature/stress conditions for 750 to 1000 hours as shown in Table 1. Stepped-temperature stress-rupture (STSR) procedures from  $800^{\circ}$ C were used in  $100^{\circ}$ C/24-hour increments prior to isothermal testing at temperatures of  $1000^{\circ}$ C and  $1200^{\circ}$ C. Specimens surviving the stress-rupture test for 750 to 1000 hours were unloaded, cooled, and removed from the furnace for measurements of weight change, permanent strain, and retained RT MOR, also shown in Table 1.

Stress Rupture (Temp. <sup>O</sup> C)	Time (hr)	ε(hr-1) (% Perm. Strain)	Oxidation Rate Const. (kg <sup>2</sup> m <sup>-4</sup> s <sup>-1</sup> )	RT MOR After Stress- Rupture Test
700	750	n.m.	n.m.	570
1000	750	n.m.	$2.0 \times 10^{-13}$	390-510
1200	1000	1.0-2 x 10-6	$6.5 \times 10^{-13}$	486-670

Table 1. CREEP, OXIDATION, AND MOR DATA

(STSR from 800°C prior to all tests at 1000°C and 1200°C)

 $\sigma$  = 300 MPa

Measurements of weight gain on stress-rupture bars indicated that they have good oxidation resistance. Oxidation rate values calculated for specimens surviving at  $1000^{\circ}\text{C}$  and  $1200^{\circ}\text{C}$  were  $2.0 \times 10^{-13}$  and  $6.5 \times 10^{-13} \text{ kg}^2\text{m}^{-4}\text{s}^{-1}$ , respectively. Essentially no change in weight was noted at  $700^{\circ}\text{C}$  within the limits and accuracy of measurement. Specimens appeared uniform in surface color with a very thin, coherent oxidation layer.

The amount of permanent strain in each stress-rupture bar was determined by measuring bar curvature under a microscope and by a photographic technique. Within the limits of measurement, bars tested at  $1200^{\circ}$ C for 1000 hours exhibited 0.1 to 0.2% permanent strain. An upper limit strain rate of 1.5 to 2.0 x  $10^{-6}$ /hr was calculated from these measurements. It is regarded as an upper limit because the measurement includes both the primary and secondary (steady state) creep stages. Permanent strain values were not measurable for specimens tested at  $700^{\circ}$ C and  $1000^{\circ}$ C (values less than 0.1%).

Retained RT MOR values are shown in Table I for stress-rupture survivors. Values range from 390 to 670 MPa and are primarily strength limited by residual pores as observed at the fracture origins (Figure 4). Occasionally, an agglomerate, as shown in Figure 5 was observed at the origin of failure.

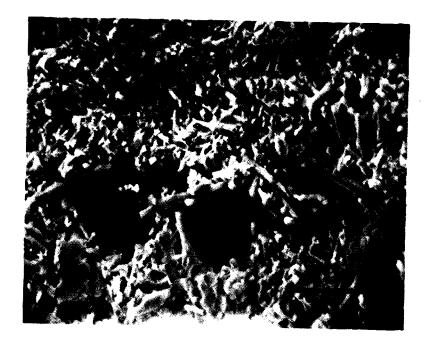


Figure 4. Pore cluster at fracture origin of sintered specimen.

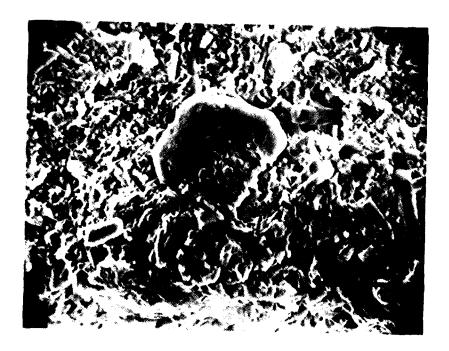


Figure 5. Agglomerate at fracture origin of sintered specimen.

#### **SUMMARY**

Reaction-bonded Si<sub>3</sub>N<sub>4</sub> bars containing  $6\%Y_2O_3$  and  $1\%Fe_2O_3$  were sintered to high density (>97%) using high N<sub>2</sub> pressures (up to 8.0 MPa) and temperatures of  $1875^{\circ}C$  to  $2000^{\circ}C$ . Bars which were oxidized before sintering had higher sintered densities than those which were not preheat treated. The development of more silica-rich compositions resulted in the formation of yttrium-silicate phases which appear to promote densification for the sintering parameters used in this study.

Stress-rupture tests on sintered bars (with and without preoxidation) show no intermediate temperature (700°C to 1000°C) degradation under stress (300 MPa) for times of 750 to 1000 hours. The bars exhibited good oxidation resistance to 120°C temperatures and low residual permanent strain was noted after 1000 hours at 1200°C. Retained MOR values after stress-rupture exposure at 700°C, 1000°C, and 1200°C were similar to as-sintered material values. Strength-limiting defects were usually residual pores and pore clusters.

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## **ACKNOWLEDGMENT**

This research was sponsored by the Advanced Materials Development Program, Office of Transportation Systems, U.S. Department of Energy/MTL IEA DE-A105-21411 with DOE's Oak Ridge Operations. The author thanks J. Smyth, Garrett Turbine Engine Company, Phoenix, AZ, and K. Styhr, AiResearch Casting Company, Torrance, CA, for supplying the doped RBSN bars; P. Noonan and T. Stefanick, MTL, for stress-rupture testing and permanent strain measurements.

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Predoped RBSN bars containing 6.7203 and 1%Fe<sub>2</sub>03 were sintered to high density using a two-step pressure/temperature cycle. Some bars were preoxidized at 100000 prior to sintering to increase the oxygen content (5102) of the specimens and to determine its effect to sintering behavior. Modulus of rupture and stress-rupture data were obtained on both as-received and heat-treated specimens. Predoped RBSN bars containing 6%Y<sub>2</sub>O<sub>3</sub> and 1%Fe<sub>2</sub>O<sub>3</sub> were sintered to high density using a two-step pressure/temperature cycle. Some bars were preoxidized at 10000°C prior to sintering to increase the oxygen content (SiO<sub>2</sub>) of the specimens and to determine its effect on sintering behavior. Modulus of rupture and stress-rupture data were obtained on both as-received and heat-treated specimens. UNCLASSIFIED UNLIMITED DISTRIBUTION UNCLASSIFIED UNLIMITED DISTRIBUTION Silicon nitrides Silicon nitrides Key Words Key Words Sintering Oxidation Sintering Oxidation 용 8 Technical Report MTL TR 86-1, January 1986, 8 pp - illus-tbl, Interagency Agreement DE-A105-21411 Technical Report MTL TR 86-1, January 1986, 8 pp -illus-tbl, Interagency Agreement DE-A105-21411 Army Materials Technology Laboratory Watertown, Massachusetts 02172-0001 EFFECT OF OXIDATION ON THE DENSIFICA-TION OF SINTERABLE RBSN -Army Materials Technology Laboratory Watertown, Massachusetts 02172-0001 EFFECT OF OXIDATION ON THE DENSIFICA-TION OF SINTERABLE RBSN -George E. Gazza George E. Gazza U.S. U.S. Predoped RBSN bars containing 6%Y<sub>2</sub>O<sub>3</sub> and 1%Fe<sub>2</sub>O<sub>3</sub> were sintered to high density using a two-step pressure/temperature cycle. Some bars were preoxidized at 10000C prior to sintering to increase the oxygen content (SiO<sub>2</sub>) of the specimens and to determine its effect on sintering behavior. Modulus of rupture and stress-rupture data were obtained on both as-received and heat-treated specimens. Predoped RBSN bars containing 6%Y<sub>2</sub>0<sub>3</sub> and 1%Fe<sub>2</sub>0<sub>3</sub> were sintered to high density using a two-step pressure/temperature cycle. Some bars were preoxidized at 10000<sub>0</sub> prior to sintering to increase the oxygen content (510<sub>2</sub>) of the specimens and to determine its effect on sintering behavior. Modulus of rupture and stress-rupture data were obtained on both as-received and heat-treated specimens. UNCLASSIFIED UNLIMITED DISTRIBUTION UNCLASSIFIED
UNLIMITED DISTRIBUTION Sintering Silicon nitrides Sintering Silicon nitrides Key Words **Key Words** Oxidation Oxidation B 8 Technical Report MTL TR 86-1, January 1986, 8 pp - illus-tbl, Interagency Agreement DE-A105-21411 Technical Report MTL TR 86-1, January 1986, 8 pp - illus-tbl, Interagency Agreement DE-Al05-21411 Army Materials Technology Laboratory Watertown, Massachusetts 02172-0001 EFFECT OF OXIDATION ON THE DENSIFICA. Army Materials Technology Laboratory Wateriamn. Massachusetts 02172-0001 Watertown, Massachusetts 02172-0001 EFFECT OF OXIDATION ON THE DENSIFICA TION OF SINTERABLE RBSN -TION OF SINTERABLE RBSN -George E. Gazza George E. Gazza u.s. u.s.

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